

BACHELOR OF SCIENCE (CBCS-2018 COURSE)
T. Y. B. Sc. Sem-V : WINTER- 2022
SUBJECT : CHEMISTRY : INORGANIC CHEMISTRY-I

Day : Friday

Time : 02:00 PM-05:00 PM

Date : 9/12/2022

W-18415-2022

Max. Marks : 60

N.B.:

- 1) All questions are **COMPULSORY**.
- 2) Figures to the right indicate **FULL** marks.

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- Q.1** Attempt **ANY TWO** of the following: [12]
- a) Discuss postulates of Valence Bond Theory.
 - b) Write IUPAC Nomenclature for following compounds:
i) $[\text{Co}(\text{NH}_3)(\text{NO}_2)_3]$ ii) $\text{K}_4[\text{Ni}(\text{CN})_4]$ iii) $[\text{Cr}(\text{en})_3]\text{Cl}_3$
 - c) Write a note on 'Immersed Corrosion'.
- Q.2** Attempt **ANY TWO** of the following: [12]
- a) What is geometrical isomerism in coordination compounds? Explain following types of geometrical isomers with suitable examples.
i) $[\text{M}(\text{A}-\text{A})_2\text{XY}]$ ii) $[\text{M}\text{A}_3\text{X}_3]$
 - b) Discuss postulates of Crystal Field Theory.
 - c) Write limitations of Valence Bond Theory.
- Q.3** Attempt **ANY TWO** of the following: [12]
- a) Explain splitting of d-orbitals in a Tetrahedral complex on the basis of Crystal Field Theory.
 - b) Explain following types of structural isomerism with suitable examples:
i) Linkage isomerism ii) Coordination isomerism
 - c) Write a note on : Atmospheric Corrosion
- Q.4** Attempt **ANY THREE** of the following: [12]
- a) Write a comparison between Metal Complex and Metal Chelate.
 - b) Explain bonding and geometry in following complexes on the basis of VBT:
 $[\text{Cu}(\text{NH}_3)_4]^{+2}$ ii) $[\text{Mn}\text{Br}_4]^{-2}$
 - c) Discuss how purity of the metal and pH of medium affects corrosion.
 - d) Calculate CFSE for following systems in a strong octahedral field:
i) d^5 system ii) d^4 system
- Q.5** Attempt **ANY FOUR** of the following: [12]
- a) Explain Sidgwick model for Effective Atomic Number with suitable example.
 - b) How nature of ligand and charge on the metal ion affect 10Dq of the complexes?
 - c) State the postulates of Werner's coordination theory.
 - d) Draw optical isomers for following type of complex with suitable example:
 $[\text{M}\text{A}_2\text{X}_2\text{Y}_2]$
 - e) Discuss any three methods for prevention of corrosion.
 - f) What are merits and demerits of Sidgwick theory of complexes?

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