BACHELOR OF SCIENCE (CBCS-2018 COURSE) T. Y. B. Sc. Sem-VI :SUMMER- 2022 SUBJECT: CHEMISTRY: PHYSICAL CHEMISTRY-II

Day: Saturday Time: 11:00 AM-02:00 PM Date: 2/7/2022 S-18470-2022 Max. Marks: 60 N.B. All questions are **COMPULSORY**. 1) Figures to the **RIGHT** indicate **FULL** marks. 2) Draw neat and labeled diagram wherever necessary. 3) Q.1 Attempt ANY TWO of the following: (12)Sketch and label the molecular energy level diagram. Define the terms: wavelength, frequency, velocity and wave number. b) Elaborate on fundamental equation of molecular spectroscopy. c) Attempt ANY TWO of the following: (12)0.2 Explain vapour temperature method for the determination dipole moment. Discuss use of dipole moment in determine of molecular structure. b) Give the list of different types of crystal systems. c) Attempt ANY TWO of the following: (12)Derive Bragg's condition for maximum reflection. Explain experimental setup for the study of Raman Spectroscopy. b) Elaborate on refractive index. c) Attempt ANY THREE of the following: (12)0.4 Calculate the frequency and wavenumber associated with radiation of wavelength 400 m μ . If the bond length of ¹H³⁵Cl is 1.274 Å, calculate reduced mass and moment b) of inertia of the molecule. (N= 6.023×10^{23}) The half-life period of radium is 1580 years. Calculate its disintegration c) constant. A crystal plane intercepts the three rectangular axes at the multiples of unit d) distance 1, 3/2 and 2. What will be the Miller indices of the plane? Attempt ANY FOUR of the following: (12)0.5 Explain types of β –decay. Obtain the expression for decay constant. b) Sketch and label the Raman spectrum. c) Draw diagrams of (100), (110) and (111) planes of simple cubic system. d) Explain the Laue method for determination of crystal structure. e) Draw diagrams of simple cubic lattice, face centered cubic lattice and body

f)

entered cubit lattice.