

**BACHELOR OF SCIENCE (CBCS-2018 COURSE)**  
**T. Y. B. Sc. Sem-VI :SUMMER- 2022**  
**SUBJECT : PHYSICS : THERMODYNAMICS & STATISTICAL PHYSICS**

Day : Tuesday  
Date : 12/7/2022

**S-18466-2022**

Time : 11:00 AM-02:00 PM  
Max. Marks : 60

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**N.B.:**

- 1) All questions are **COMPULSORY**.
  - 2) Figures to the **RIGHT** indicate full marks.
  - 3) Draw neat labeled diagrams **WHEREVER** necessary.
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**Q. 1** Attempt any **Two** of the following. **(12)**

- (a) Explain the fundamental assumptions of Kinetic theory of gas.
- (b) Derive an expression for the coefficient of viscosity of gas ( $\eta$ ) in terms of mean free path.
- (c) Determine the mean free path of the molecule of hydrogen at NTP, given that density of hydrogen is  $8.96 \times 10^{-5} \text{ g cm}^{-3}$ , coefficient of viscosity  $8.6 \times 10^{-5}$  C.G.S. units and  $k = 1.38 \times 10^{-16} \text{ ergs } ^\circ\text{K}^{-1}$ .

**Q. 2** Attempt any **Two** of the following. **(12)**

- (a) Derive and explain the Maxwell's Relation.
- (b) With neat suitable diagram explain the porous plug experiment.
- (c) Show that for an ideal gas  $C_p - C_v = R$  and for a real gas obeying Vander Waals equation  $C_p - C_v = R\{1 + 2a/RTV\}$  approximately.

**Q. 3** Attempt any **Two** of the following. **(12)**

- (a) Derive binomial expression for random walk problem.
- (b) Explain the thermal interaction mechanism.
- (c) Describe Gibbs and Helmholtz function.

**Q. 4** Attempt any **Three** of the following. **(12)**

- (a) Calculate the change in entropy when 10 grams of ice at  $0^\circ\text{C}$  is converted into water at the same temperature. (Given: Latent heat of ice = 80 cal/gram).
- (b) Explain air liquefier with suitable diagram.
- (c) Explain the behavior of density of states of system
- (d) Write a short note on ensemble.

**Q. 5** Attempt any **Four** of the following. **(12)**

- (a) Describe the probability distribution function.
- (b) The diameter of molecule of a gas is  $2.3 \times 10^{-10} \text{ m}$ . The mean free path is  $2.05 \times 10^{-7} \text{ m}$ . Calculate the number of molecules of the gas per c.c.
- (c) Explain the terms: i) Entropy ii) Enthalpy
- (d) Explain the variation of  $C_v$  with pressure.
- (e) Write a short note on transport phenomena.
- (f) Describe the terms canonical and micro canonical ensemble.

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