BACHELOR OF SCIENCE (CBCS - 2016 COURSE) S. Y. B Sc. Sem-IV :SUMMER- 2022

SUBJECT: CHEMISTRY: ORGANIC & INORGANIC CHEMISTRY-IV

Day: Friday Time: 03:00 PM-06:00 PM Date: 8/7/2022 S-14817-2022 Max. Marks: 60 N.B. 1) All questions are COMPULSORY 2) Figures to the right indicates FULL marks 3) Answers to the both sections should be written in SAME answer book SECTION - I Q.1 Attempt **ANY TWO** of the following: (12)Describe Killiani-Fischer synthesis of aldotriose. Write the mechanism for nitration of nitrobenzene with mixed acid. Comment on stability of arenium ion of it. c) What is Green chemistry? Write in detail about green reagents. 0.2 Attempt ANY THREE of the following: (12)How will you obtain following compounds from Aniline a) i) Benzene ii) Phenol Give action of nitrous acid on primary and secondary aromatic amines. b) How will you effect following conversions i) Methyl alcohol to ethyl alcohol ii) Methyl iodide to ethyl alcohol What are the reactions of following reagents on glucose i) Br₂ Water ii) dil. HNO₃ Attempt **ANY ONE** of the following: (06)Q.3 A)Discuss the importance of Microwave technique in green reactions with a) examples. Explain any six principles of green chemistry with suitable examples. b) **SECTION-II** Attempt ANY ONE of the following: (06)Q.3 B) Explain the Lewis concept of acids and bases. What are its assumptions? a) What are its merits and demerits? What is hydrogen bonding? Explain the effect of hydrogen bonding on physical state of the compound. **Q.4** Attempt ANY TWO of the following: (12)Explain the Arrhenius Concept of acids and bases. Write its merits and Explain physiological role of Potassium, Nitrogen and Phosphorus. b) Explain effect of hydrogen bonding on density of compounds in detail. Attempt ANY FOUR of the following: (12)Q.5

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What is the origin of Van-der-Waal's forces?

Write a short note on 'Hydracids'.

Explain in brief polymers containing Phosphorus.

Discuss in brief solvent system of acids and bases.

Explain in detail about trends in the strength of oxyacids.

a)

c)

d)

e)