BACHELOR OF TECHNOLOGY (C.B.C.S.) (2021-COURSE) B. Tech. Sem - II E&TC: SUMMER- 2022 **SUBJECT: CHEMISTRY OF ELECTRONIC MATERIALS**

Time: 10:00 AM-01:00 PM

Day: Thursday S-24100-2022 Max. Marks: 60 Date: 28-07-2022 N.B.: All questions are **COMPUSLORY**. 1) Figures to the right indicate FULL marks. 2) Draw neat and labeled diagram WHEREVER necessary. 3) Assume suitable data if necessary 4) Explain the different types of glasses and their applications. [10] **Q.1 OR** Explain the various factors affecting the electrical resistivity of materials. [10] Q.1 What are compound semiconductors? Briefly explain its composition, **Q.2** [10] properties and applications. OR **Q.2** Explain how piezoelectricity is generated in materials with suitable examples. [10] Explain the seven crystal systems with suitable examples. Q.3 [10] OR Q.3 Briefly explain: [10] Molecular crystals a) b) Ionic crystals Covalent crystals c) Explain the preparation, properties and applications of two important [10] Q.4 conductive polymers. OR By taking an example briefly explain the requirements in a polymer to act as [10] Q.4 conducting polymer. What are reference electrodes? With a neat diagram explain the working of [10] Q.5 calomel electrode and standard hydrogen electrode. OR Explain the construction and working of dry cell and Ni-Cd batteries. Q.5 [10] Explain the instrumentation and applications of IR spectroscopy. [10] **Q.6** OR **Q.6** Derive combined beer-Lambert's law. What is molar absorptivity? [10]