

**BACHELOR OF TECHNOLOGY (C.B.C.S.) (2014 COURSE)**

**B.Tech.Sem - VI CHEMICAL : : SUMMER - 2022**

**SUBJECT : SEPARATION TECHNIQUES**

Day : Monday  
Date : 13-06-2022

**S-13508-2022**

Time : 02:30 PM-05:30 PM  
Max. Marks : 60

**N.B.**

- 1) All questions are **COMPULSORY**.
- 2) Figures to the **RIGHT** indicate **FULL** marks.
- 3) Use of non-programmable calculator is **allowed**.
- 4) Assume suitable data **WHEREVER** necessary.

- Q.1 a)** Describe positive deviation from ideality with suitable diagram. **(05)**  
**b)** What is azeotropic distillation? Describe its industrial applications. **(05)**

**OR**

- Q.1** An equimolar feed mixture of 100 moles containing A and B is differentially distilled such that residue contains 23% of A. Estimate the composition of distillate and residue **(10)**

X	0	1	8	14	21	29	37	46	56	66	97	100
Y	0	3	16	28	39	50	59	65	76	83	99	100

- Q.2** Derive q-line equation and show that,  $y = \frac{q}{q-1}x - \frac{xf}{(q-1)}$ . **(10)**

**OR**

- Q.2** It is desired to separate a mixture of 50% vapor and 50% liquid in a plate type distillation column. The feed contains 45 mole % A and the top product is to contain 96 mole % of A. **(10)**

A bottom product is to contain 5 mole % A. Determine the minimum reflux ratio and the number of theoretical plates needed if a reflux ratio of twice the minimum is used.

The equilibrium data is as below

x	0	0.1	0.16	0.3	0.4	0.5	0.6	0.7	0.8	0.9	1.0
y	0	0.215	0.3	0.52	0.625	0.725	0.78	0.89	0.89	0.95	1.0

x and y are the mole fractions of A in liquid and vapour phase respectively.

- Q.3** Draw schematic arrangement of multistage cross current adsorption operation and write a material balance of solute for stage 1 and 2. Represent the operation graphically and describe the steps involved in the determination of no. of stages needed. **(10)**

**OR**

- Q.3** 500 Kg/min of dry air at 20°C and carrying 5 kg of water vapor/min is to be dehumidified with silica gel to obtain air with 0.001 kg of water vapour/kg of dry air. The adsorption operation has to be carried out counter-currently with 25 kg/min of dry silica gel. How many theoretical stages are required and what will be water content in the silica gel leaving the last stage? **(10)**

$\frac{\text{kg of water vap}}{\text{kg of dry silica gel}}, X$  0 0.05 0.10 0.15 0.20

$\frac{\text{kg of water vap}}{\text{kg of dry air}}, Y$  0 0.0018 0.0036 0.005 0.0062

- Q.4** Draw a multistage cross current extraction operation for an insoluble system and make a mass balance about stage n. describe the graphical determination of no. of stages in the cross-current operation. **(10)**

**OR**

- Q.4** 1000 kg per hour of a solution of C in A containing 20% C by weight is to be counter currently extracted with 400 kg/h of solvent B. the component A and B are insoluble. **(10)**

wt. of C / wt. of A	0.05	0.20	0.3	0.45	0.5	0.54
wt. of C / wt. of B	0.25	0.4	0.5	0.65	0.7	0.74

How many theoretical stages will be required to reduce the concentration of C to 5 % in effluent?

- Q.5** Describe with neat sketch different types of equilibrium diagrams encountered in leaching operation. **(10)**

**OR**

- Q.5** Describe various steps involved in leaching operation. What factors have direct impact on the success of the leaching operation? Describe these factors in detail. **(10)**

- Q.6** What are the limitations and industrial applications of i) Ultrafiltration ii) Nano-filtration? **(10)**

**OR**

- Q.6** What is Reverse Osmosis? What are its advantages over other membrane separation processes? State its industrial applications. **(10)**

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