MASTER OF SCIENCE (CHEMISTRY) (CBCS - 2018 COURSE) M.Sc. (Chemistry) Sem-IV AC :SUMMER- 2022 SUBJECT : ADVANCED ANALYTICAL TECHNIQUES

Day: Saturday Time: 03:00 PM-06:00 PM Date: 02-07-2022 S-20179-2022 Max. Marks: 60 $\overline{\text{N.B.}}$ 1) All questions are **COMPULSORY**. 2) Figures to the right indicate FULL marks. Draw neat and labelled diagram WHEREVER necessary. 3) 4) Use of log table / non – programmable scientific calculator is **ALLOWED**. SECTION - I Q.1 Attempt ANY THREE of the following: (15)a) Discuss liquid phase chemiluminescent titrations. Give an account of applications of Turbidometry and Nephelometry in detail. b) Explain MID IR reflection spectrometry with suitable diagram. c) Explain theory of Raman spectroscopy. d) Explain in detail the qualitative and quantitative analysis of IR spectroscopy. e) 0.2 A) Attempt **ANY TWO** of the following: (10)Describe fluorimetry apparatus in brief with suitable sketch. i) Explain any two detectors used in IR instruments. ii) iii) Explain the stretching and bending vibrations of a polyatomic molecular. B) Solve **ANY ONE** of the following: (05)What is the energy possessed by a radiation of wave number 1050 cm⁻¹? i) The fundamental frequency of C = O is 2075 cm⁻¹, then calculate the force ii) constant of this bond. (Given: Atomic weights - C = 12.05, O = 15. 99, $c = 3 \times 10^{10}$ cm/s) **SECTION - II** Q.3 Attempt ANY THREE of the following: (15)Explain the interaction of X – rays with matter. a) Explain FT – NMR spectrometer with suitable sketch and its applications in b) various industries. Give a sketch of electron microscopy with diagram and explain applications c) d) Why ESCA give satellite peak? Explain all the possible reasons in detail. Discuss the advantages and disadvantages of X – ray fluorescence method. e) Q.4 A) Attempt ANY TWO of the following: (10)Describe Auger electron spectroscopy with suitable diagram. Describe the construction & working of Coolidge tube which is used for ii) production of X – rays. Explain the use of MRI in medicinal chemistry. iii) Solve **ANY ONE** of the following: (05)B) Calculate the frequency of ESR line for an unpaired electron at magnetic flux i) density of 0.3572 T. (Given: g = 2.16, $n = 6.626 \times 10^{-34}$ Js) Calculate the NMR absorption frequency in Hz for ⁹F nucleus in a magnetic ii) fields. (Given: $\lambda = 2.52 \times 10^8 \text{ rad } \text{T}^{-1} \text{ s}^{-1}$)