BACHELOR OF SCIENCE (CBCS - 2016 COURSE) T. Y. B Sc. Sem-VI :SUMMER- 2022

SUBJECT: CHEMISTRY: INORGANIC CHEMISTRY-II

Time: 11:00 AM-02:00 PM Day: Tuesday S-15051-2022 Max. Marks: 60 Date: 5/7/2022 N.B.: All questions are **COMPULSORY**. 1) 2) Figures to the right indicate FULL marks. 0.1 Attempt **ANY TWO** of the following: [12] a) Construct M.O energy level diagram for CO molecule and discuss the bonding Silicon dopped with Gallium shows P-type semi-conductivity. Explain with diagram. c) Define homogeneous catalysis and write different types and properties of homogeneous catalysts. **Q.2** Attempt **ANY TWO** of the following: [12] a) Construct M.O energy level diagram for F₂ molecule and discuss the bonding **b)** Write a comparison between VBT and MOT. c) What is catalysis? Write the mechanism of polymerization by Ziegler-Natta Catalyst. 0.3 Attempt **ANY TWO** of the following: [12] a) Write a classification of Toxic substances. b) Describe ion-exchange method for the separation of the lanthanide elements. c) What is Intrinsic semi-conductivity? Why pure graphite shows intrinsic semiconductivity? Q.4 Attempt **ANY THREE** of the following: [12] a) ZnO when heated shows n-type semi-conductivity. Explain with diagram. **b)** Explain any one method for preparation of Transuranic elements. c) Explain impact of Toxic chemicals on Enzymes. **d)** Write the rules of LCAO on the basis of M.O theory. Attempt **ANY FOUR** of the following: Q.5 [12] a) Explain applications of lanthanides. b) Write the total reaction of Monsanto process and write the advantages of the catalyst used in the process. c) Explain in brief toxic effects of Arsenic. **d)** Why monovalent metals exhibit good electrical conductivity?

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e) Write any three points of differences between B.M.O. and A.B.M.O.

f) What is Lanthanide contraction?