M. Sc. (Environment Science and Technology) Sem - I (CBCS) (2013 Course): WINTER - 2018 SUBJECT: ENVIRONMENTAL CHEMISTRY

Day: Monday Time: 10.00 AM TO 01.00 PM W-2018-1226 Date: 19/11/2018 Max. Marks: 60 N.B.: 1) Answer any **FIVE** questions. 2) Figures to the right indicate FULL marks. Q.1 a) What are the enzymes? Write the mechanism of enzyme catalyzed reactions. (06)b) Explain the formation of PAN during smog. (06)Q.2 a) Explain the acid-base and ion exchange reactions in soil. (06)b) What are alcohols? How are monohydric alcohols classified? Write any two (06)methods of synthesis of ethyl alcohol. Q.3 a) Define corrosion. Describe the types of corrosion (06)**b)** Define any two of the following: (06)i) Le- Chattelier's principle ii) Activity coefficient iii) Coagulation **O.4** a) What is EL- Nino? How does it affect the global climate? (06)b) Give a brief account on variations in equilibrium relationships. (06)(06)Q.5 a) Describe construction and working of HPLC. **b)** Write short notes on any two of the following: (06)i) Buffer index ii) Semiconductors iii) Fluoridation Q.6 a) Comment on anomalous behavior of water. (06)b) Discuss the reactions of free radicals in the atmosphere. (06)

* * *