## FINAL YEAR B.PHARM. SEMESTER-VII (2011 COURSE) : SUMMER - 2018

## SUBJECT : PHARMACEUTICAL ANALYSIS – V

| Day<br>Date |                | <b>S-2018-3977</b> Time: Max. Ma  | <b>02.00 PM TO 05.00 P</b> I<br>rks : 80 |  |  |
|-------------|----------------|---|--|--|--|
| N.B.        | 1)<br>2)<br>3) | Q.1 and Q.5 are <b>COMPULSORY</b> . Out of the remaining atte questions from Section – I and any <b>TWO</b> questions from Sections should be written in <b>SEPARATI</b> Figures to the right indicate <b>FULL</b> marks.   | estions from Section – II.               |  |  |
|             |                | SECTION – I   |  |  |  |
| Q.1         |                | Answer any FIVE of the following  | (10)                                     |  |  |
|             | a) b) c) d)    | Define wave number and frequency. Explain concept of Instrumental Analysis. Define the term chromophore. Differentiate the following structures using Woodward Fiser Explain 'Holographic gratings. List out properties of molecules used for instrumental analys |  |  |  |
| Q.2         | a)             | Write the effect of conjugation on λmax, explain with examp   | oles. (08)                               |  |  |
|             | b)             | Predict $\lambda$ max for the following structure.  | (07)                                     |  |  |
| Q.3         | a)             | Explain the construction working and advantages of PMT.   | (08)                                     |  |  |
|             | b)             | List out dispersive elements used in spectral instrumentation describe in detail 'gratings'.  | and (07)                                 |  |  |
| Q.4         |                | Write short notes on any <b>THREE</b> of the following:   | (15)                                     |  |  |
|             | a)<br>b)       | Woodward Fisers rule for enones.  Classification of instrumental methods of analysis with type and molecular interactions.  | es of atomic                             |  |  |
|             | c)<br>d)       | Raman spectroscopy Quantitative analysis by UV spectroscopy.  | P.T.O.                                   |  |  |

## SECTION – II

| Q.5         |   | Answer any FIVE of the following  | (10) |  |
|-------------|---|---|------|--|
|             | a)<br>b)<br>c)  | Write the basic requirement of a molecule to be IR active. List out factors affecting fluorescence. Calculate the vibrational degrees of freedom of a linear molecule with 10 atoms.          |      |  |
|             | d)  | <ul> <li>d) Explain the principle of phosphorimetry.</li> <li>e) List out the detectors used in IR spectroscopy.</li> <li>f) Advantages of Raman spectroscopy over IR spectroscopy</li> </ul> |      |  |
|             | e)  |   |      |  |
|             | f)  |   |      |  |
| Q.6         | a)  | How IR spectroscopy is useful to distinguish following structures   | (08) |  |
|             | i)  | CH <sub>3</sub> -CH <sub>2</sub> -CH <sub>2</sub> -CH <sub>2</sub> OH CH <sub>3</sub> -CH <sub>2</sub> -CH <sub>2</sub> -C-H  |      |  |
|             | ii)   | $CH_2$ = $CH$ - $CH$ = $CH_2$ $CH_3$ - $CH_2$ - $CH_3$  |      |  |
|             | b)  | Write a note on molecular vibrations  |      |  |
| <b>Q.</b> 7 |   | Explain in details instrumentation, advantages and disadvantages of fluorimetry   |      |  |
| Q.8         |   | Write short notes on any THREE of the following:  |      |  |
|             | <ul> <li>a) Principle and applications of turbidimetry</li> <li>b) Necessary conditions for quantitation by nepheloturbidometry</li> <li>c) Sampling methods by IR spectroscopy</li> <li>d) Compare nephelometry, turbidometry and UV spectrometry</li> </ul> |   |      |  |

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