

S.Y.B.PHARM. SEMESTER-IV (CBCS - 2015 COURSE) :
SUMMER - 2018

SUBJECT : PHARMACEUTICAL ANALYSIS - II

Day : **Saturday**
Date : **28/04/2018**

S-2018-3923

Time : **02.00 PM TO 05.00 PM**
Max. Marks : 60

N. B. :

- 1) Question No – 1 and 5 are **COMPULSORY**. Out of remaining questions attempt **any Two** from section - I and **any Two** questions from section - II
 - 2) Answers to both the sections should be written in **SEPARATE** answer books.
 - 3) Figures to the right indicate **FULL** marks.
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SECTION - I

- Q.1** Attempt **ANY FIVE** of the following: **(10)**
- a) What is differentiating pulse polarography
 - b) Classify various electro analytical techniques
 - c) What is acid error and alkaline error in potentiometry?
 - d) What is 'maxima' in polarography? How is it reduced?
 - e) With what is calibration of pH electrode done?
 - f) Why is it necessary to remove oxygen from polarographic equipment?
- Q.2** a) Give a detailed account of Amperometric titrations **(7)**
b) Explain the principle involved in biamperometry **(3)**
- Q.3** a) Draw and explain construction and working of DME **(7)**
b) Give Ilkovic equation and its significance **(3)**
- Q.4** Write short notes on **ANY TWO** of the following: **(10)**
- a) Applications of Coulometry
 - b) Reference electrodes in potentiometry
 - c) Mass transfer by various modes

SECTION - II

- Q.5** Attempt **ANY FIVE** of the following: **(10)**
- a) Explain various factors affecting angle of refraction
 - b) Explain the term ORD and CD
 - c) How and with what is calibration of conductometer performed?
 - d) Explain the effect of dilution on molar and equivalent conductance
 - e) Give comparison between co-precipitation and post precipitation
 - f) Explain conductometric titration of Weak Acid Vs. Strong Base
- Q.6** a) Give principle, instrumentation and application of Abbe's Refractometer **(10)**
- Q.7** a) Explain the principle and instrumentation of polarimeter **(7)**
b) Give applications of Polarimetry **(3)**
- Q.8** Write short notes on **ANY TWO** of the following: **(10)**
- a) Gasometric assay of CO₂
 - b) Conductivity cell
 - c) Steps in gravimetry