

**M. SC. (ANALYTICAL CHEMISTRY) / M. SC. (ORGANIC CHEMISTRY) / M. SC. (INORGANIC CHEMISTRY) SEM-II
(CHOICE BASED CREDIT & GRADE SYSTEM) : SUMMER -**

2018

SUBJECT: FUNDAMENTALS OF ANALYTICAL CHEMISTRY

Day : **Thursday** Time : **03.00 PM TO 06.00 PM**
Date : **19/04/2018** **S-2018-0873** Max. Marks : 60

N. B. :

- 1) All questions are **COMPULSORY**.
 - 2) Figures to the right indicate **FULL** marks.
 - 3) Answers to both the sections should be written in **SEPARATE** answer book.
 - 4) Neat diagrams must be drawn **WHEREVER** necessary.
 - 5) Use of logarithmic table/ calculator is **ALLOWED**.
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SECTION - I

Q.1 Attempt **Any THREE** of the following : **(15)**

- a) Give a schematic diagram for HPLC set up and explain the function of each component.
- b) What are resins and discuss the principle of ion exchange chromatography.
- c) Explain multiple extractions are better than single extraction.
- d) Describe the principle of gas chromatography? Explain the flame ionization detector in detail.
- e) Write a note on column chromatography.

Q.2 A) Attempt **Any ONE** of the following : **(05)**

- a) Give the principle of paper chromatography and thin layer chromatography and compare the advantages and disadvantages of the two techniques.
- b) Write a note on GC-MS.

B) Solve **Any TWO** of the following : **(10)**

- a) Two litre of aqueous solution contains 32 gm of an ascorbic acid. Calculate amount of ascorbic acid extracted in ether. After
 - i) Single extraction with 100 ml ether and
 - ii) Four extractions with 50 ml ether each where $D = 10$.
- b) In chromatography separation of mixture consists of A, B, C components. Solvent front is 13.2 cm while distance traveled by A is 8.3 cm, for B 5.6 cm and for C, 7.5 cm. The unknown compound D has R_F value 0.5. Find out distance travelled by unknown compound.
- c) A GLC peak had a retention time 50 sec. The box width of the peak obtained from intersection of base line with extrapolated sides of the peak 5.5 sec. Calculate the HETP in cm/ plate. Length of column is 4 m.

P. T. O.

SECTION - II

Q.3 Attempt **Any THREE** of the following : **(15)**

- a) Explain the analysis of NO_x in air by different instrumental methods.
- b) Write a note on Ozone hole.
- c) Describe the various methods for the sampling of gaseous and vapours from air.
- d) Give an account of inorganic particulate matter and how they differ from organic particulate matter.
- e) Describe the chemistry involved in Bhopal disaster.

Q.4 Attempt **Any THREE** of the following : **(15)**

- a) Give the broad classification of water pollutants. Explain briefly.
- b) Describe the method for the estimation of the Ammonia in water sample.
- c) Define pesticide. Discuss the persistence of pesticide in detail.
- d) Explain oil pollution of water bodies.
- e) Write a note on Eutrophication.

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