

T.Y.B.SC. SEM – V (2014 COURSE) : SUMMER - 2018
SUBJECT : CHEMISTRY : ORGANIC CHEMISTRY – V (C – 53)

Day : **Monday**
Date : **16/04/2018**

S-2018-0745

Time : **03.00 PM TO 05.00 PM**
Max. Marks : 40

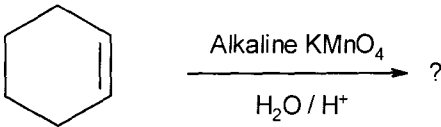
N.B.:

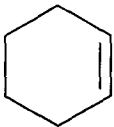
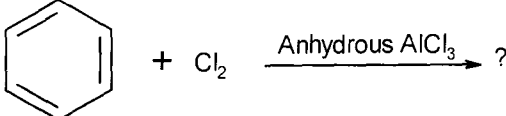
- 1) All questions are **COMPULSORY**.
- 2) Figures to the right indicate **FULL** marks.
- 3) Answers to both the sections should be written in **SAME** answer book.

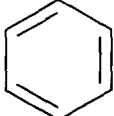
SECTION – I

- Q.1** Attempt **ANY TWO** of the following: [10]
- a) What is Sulphonation? Discuss the mechanism of sulphonation of benzene.
 - b) What is S_N^1 reaction? Discuss its mechanism. Give factors affecting on it.
 - c) Write a note on : Hydroxylation of alkene with peracids.
- Q.2** Attempt **ANY TWO** of the following: [10]
- a) Draw chair conformations of *cis* and *trans* 1, 3 – dimethyl cyclohexane and comment on their stability and optical activity.
 - b) What is E_1 reaction? Discuss its mechanism. Give factors affecting on it.
 - c) Write a note on : Activating and Deactivating groups.

SECTION – II

- Q.3** Attempt **ANY TWO** of the following: [10]
- a) What is alkylation? Discuss the mechanism of Friedel-Craft alkylation of benzene. What are its limitations?
 - b) Discuss the effect of substrate and effect of nucleophile on S_N^2 mechanism.
 - c) Write a note on ; E_1cB reaction.
- Q.4** Attempt **ANY TWO** of the following: [10]
- a) Explain the following:
 - i) Nitrobenzene on nitration gives *m* – dinitrobenzene.
 - ii) Hydration of alkene.
 - b) Predict the major product/s and suggest the mechanism:
 - i)
$$\text{CH}_3 - \overset{\text{CH}_3}{\underset{|}{\text{C}}} = \text{CH}_2 \xrightarrow{\text{O}_3, \text{Zn}/\text{H}_2\text{O}} ?$$
 - ii) 

 $\xrightarrow[\text{H}_2\text{O}/\text{H}^+]{\text{Alkaline KMnO}_4}$?
 - c) Complete the following reactions and suggest the mechanism:
 - i) $\text{CH}_3 - \text{CH} = \text{CH}_2 + \text{HBr} \xrightarrow{\text{H}_2\text{O}_2} ?$
 - ii) 

 + $\text{Cl}_2 \xrightarrow{\text{Anhydrous AlCl}_3} ?$

* * * *