

F.Y.B.SC. SEM – I (CBCS - 2016 Course) : SUMMER - 2019

SUBJECT: PHYSICS: MODERN PHYSICS

Day : Saturday
Date : 04/05/2019

S-2019-0801

Time : 11.00 A.M TO 02.00 PM
Max. Marks :60

N.B.

- 1) All questions are **COMPULSORY**.
 - 2) Figures to the **RIGHT** indicate **FULL** marks.
 - 3) Draw neat diagrams **WHEREVER** necessary.
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Q 1. Attempt any **Two** of the following. (12)

- (a) Explain energy level diagram of Hydrogen atom.
- (b) Explain the term population inversion and optical pumping in details.
- (c) Determine the energy needed to remove a neutron from the nucleus of the ${}_{28}\text{Ni}^{64}$. The mass of Ni is 63.9279 a.m.u. (Given: Mass of proton = $m_p = 1.007825$ a.m.u., Mass of neutron = $m_n = 1.008665$ a.m.u.).

Q 2. Attempt any **Two** of the following. (12)

- (a) Explain Absorption, spontaneous emission and stimulated emission in details.
- (b) Describe the construction and working of p-n junction solar cell.
- (c) Calculate input power of solar cell using the following data,
Efficiency of cell = 10%, $V_{OC} = 400$ mV, $I_{SC} = 20$ mA, Fill Factor = 0.5

Q 3. Attempt any **Two** of the following. (12)

- (a) Explain I-V characteristics of solar cell and the expression for Fill Factor (FF) for solar cell?
- (b) Draw a diagram for electromagnetic waves and explain how electromagnetic waves are produced.
- (c) Calculate the energy equivalent to 1 a.m.u.

Q 4. Attempt any **Three** of the following. (12)

- (a) Discuss in details: Renewable energy sources.
- (b) Explain Planck's hypothesis of photon.
- (c) State and explain Bohr's correspondence principle.
- (d) Write a short note on covalent bonds with example.

Q 5. Attempt any **Four** of the following. (12)

- (a) Define isotopes, isotones and isobars.
- (b) State postulate of Bohr's theory of Hydrogen atom.
- (c) Explain the term electrostatic potential and potential energy.
- (d) Describe the basic properties of nucleus.
- (e) Find the wavelength of spectral line corresponding to transition in Hydrogen atom from $n=3$ state to $n=1$ state.
- (f) Explain any three application of LASER.

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